



**GCCC**  
GULF COAST CARBON CENTER

BUREAU OF  
ECONOMIC  
GEOLOGY



TEXAS Geosciences  
The University of Texas at Austin  
Jadean School of Geosciences

# Captions to the Rescue!

Learn with Maya & Cosmo!

Let's save our planet with science & technology! Come join our CCS adventure on Earth!

Illustrators:  
Dolores van der Kolk  
Sue Hovorka  
Augela Luciano  
Frencine Mastangelo

Creators:  
Dolores van der Kolk  
Frencine Mastangelo

## Would you like to become a Carbon-not?

### Carbon-nots to the Rescue!

Join us as we commit to decreasing greenhouse gases in our atmosphere by decreasing our carbon footprint, and by supporting carbon capture and storage (CCS). Support CCS today!

Gulf Coast Carbon Center  
Bureau of Economic Geology  
The University of Texas at Austin  
To learn more, please visit us at:

[www.put-it-back.org](http://www.put-it-back.org)

This publication should be cited as:

van der Kolk, D. A., Mastrangelo, F. M., Hovorka, S. D., Tran, V. H., and Luciano, A. K., Carbon-nots to the Rescue! The University of Texas at Austin, Bureau of Economic Geology, CB2025, 12 p., doi.org/10.23867/CB2025



Cosmo and Maya are the Texas State Bird known as **Northern Mockbirds**

Mockingbirds are known to mimic a variety of sounds. Their song is a long series of phrases that they often repeat 2 to 6 times before shifting to a new sound.

They are found often in open grounds and with shrubby vegetation (hedges, fruiting bushes & thickets)

Mockingbirds eat mainly insects during the summer, but switch to eating mostly fruit in fall and winter.

**TEXAS** Geosciences  
The University of Texas at Austin  
Jackson School of Geosciences

**GCCC**  
GULF COAST CARBON CENTER



BUREAU OF  
ECONOMIC  
GEOLOGY

Texas-Louisiana Carbon Management Community

- **Population:** Stable, not endangered
  - **Clutch size:** 2 to 6 eggs (pale blue or greenish white splotched with red or brown)
  - **Incubation period:** 12 to 13 days
  - **Nesting period:** 12 to 13 days
  - **Broods:** 2 to 3
- To learn more: <https://www.allaboutbirds.org/guide/>

**What is carbon dioxide ( $\text{CO}_2$ )?**

*Cosmo,*  
I am getting too hot  
and too thirsty!  
**TOO HOT!**  
**TOO THIRSTY!**

**Oh gee Maya!**  
There is a rise in average  
temperature on Earth, because  
there is too much  $CO_2$  in the atmosphere.

**CO<sub>2</sub>** is known as a **greenhouse gas**, since this gas traps heat in our atmosphere.

$\text{CO}_2$  has been recorded in the atmosphere in studies since 1958 at the Mauna Loa Observatory in Hawaii.

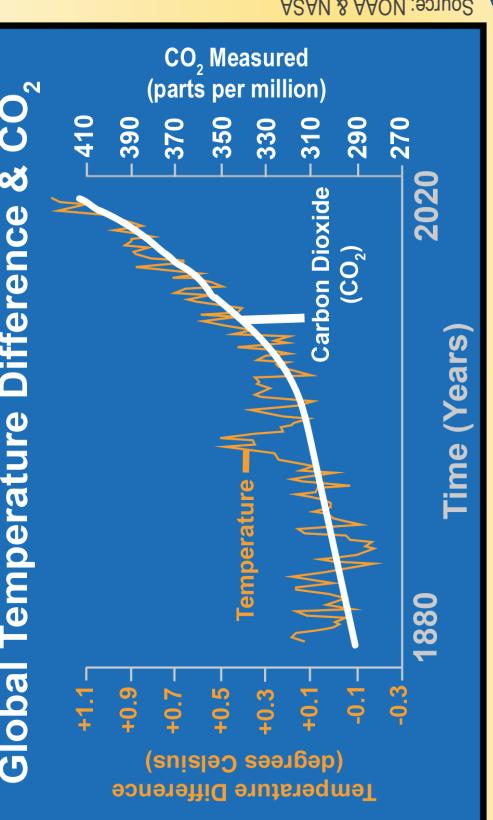
Higher concentrations of CO<sub>2</sub> are observed due to various human activities.

$\text{CO}_2$  can enter Earth's atmosphere when we burn fossil fuels like **coal**, **natural gas**, and **oil**, as well as **trees**, and **solid**

Oh gee Maya!  
There is a rise in average  
temperature on Earth, because there  
is too much  $CO_2$  in the atmosphere.  
**TOO MUCH  $CO_2$ !**

$\text{CO}_2$  comes from a variety of natural sources, but **human-related emissions** have been responsible for an increase in the atmosphere since the industrial revolution.

WHERE IS ALL  
THAT CO<sub>2</sub>  
COMING  
FROM?



Oh, dear!  
Rising heat causes  
**HOTTER HEAT WAVES,  
MORE FREQUENT  
DROUGHTS, FIRES,  
HEAVIER RAINFALL,  
and FLOODS!**

**PEOPLE,**  
can you stop  
letting  $CO_2$  go?  
**THIS has to STOP!**

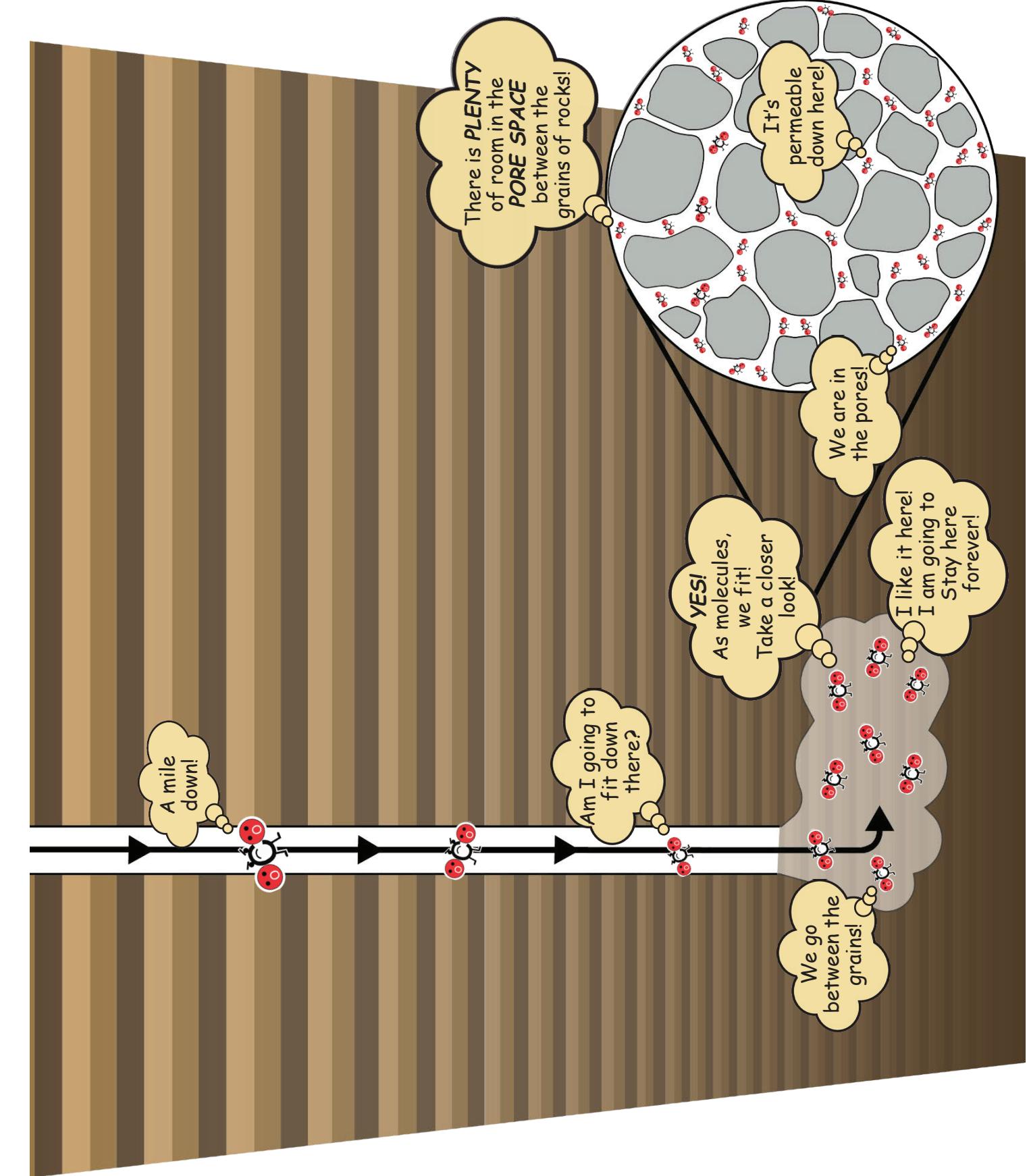
Maya,  
it is  
getting  
**HOTTER!!**

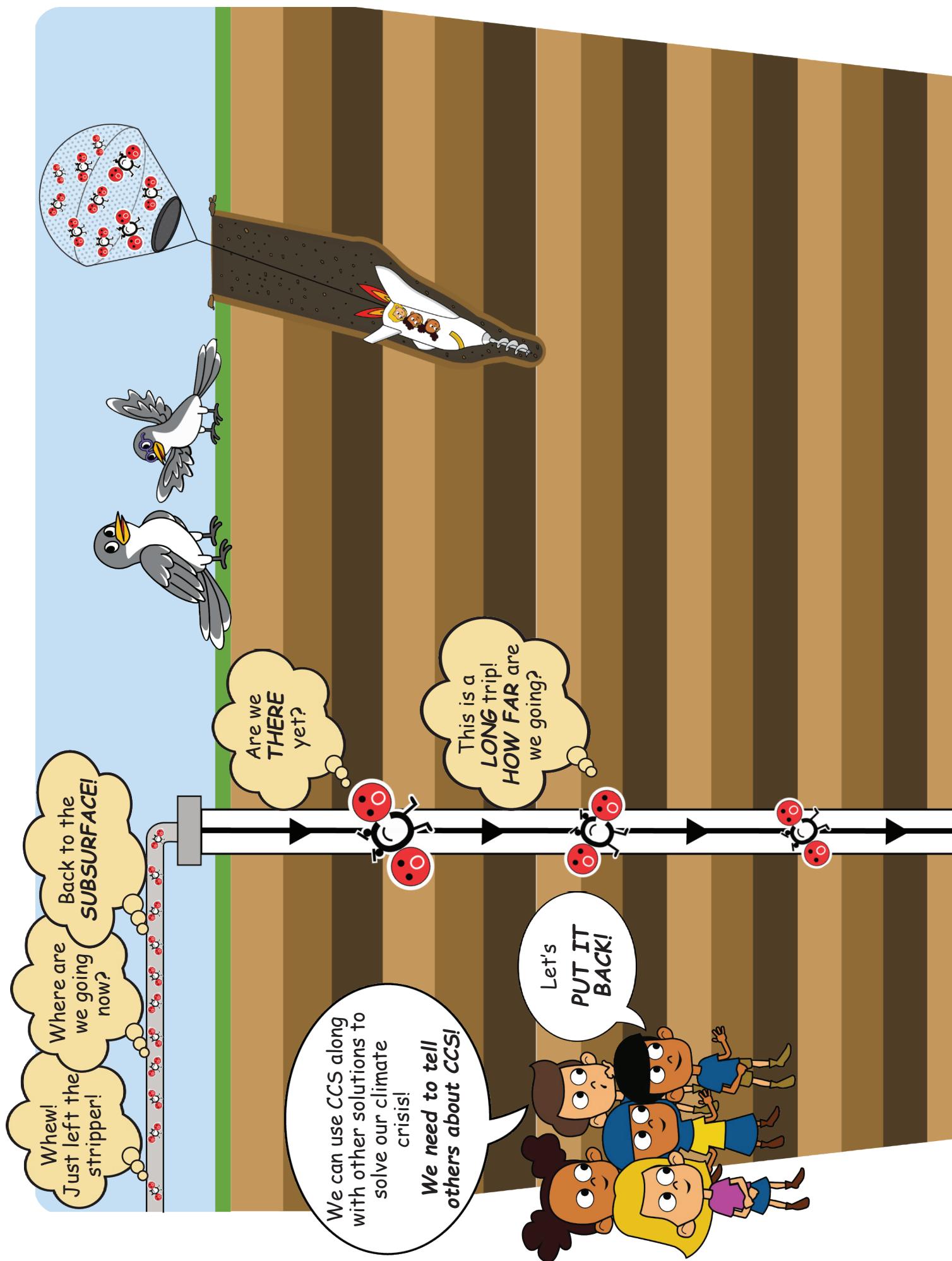
Hmm...  
WHERE IS ALL THAT CO<sub>2</sub> COMING FROM?

HOTTER!!

MORE DROUGHTS,  
DROUGHTS, FIRES,  
HEAVIER RAINFALL,  
and FLOODS!  
FLOODS!

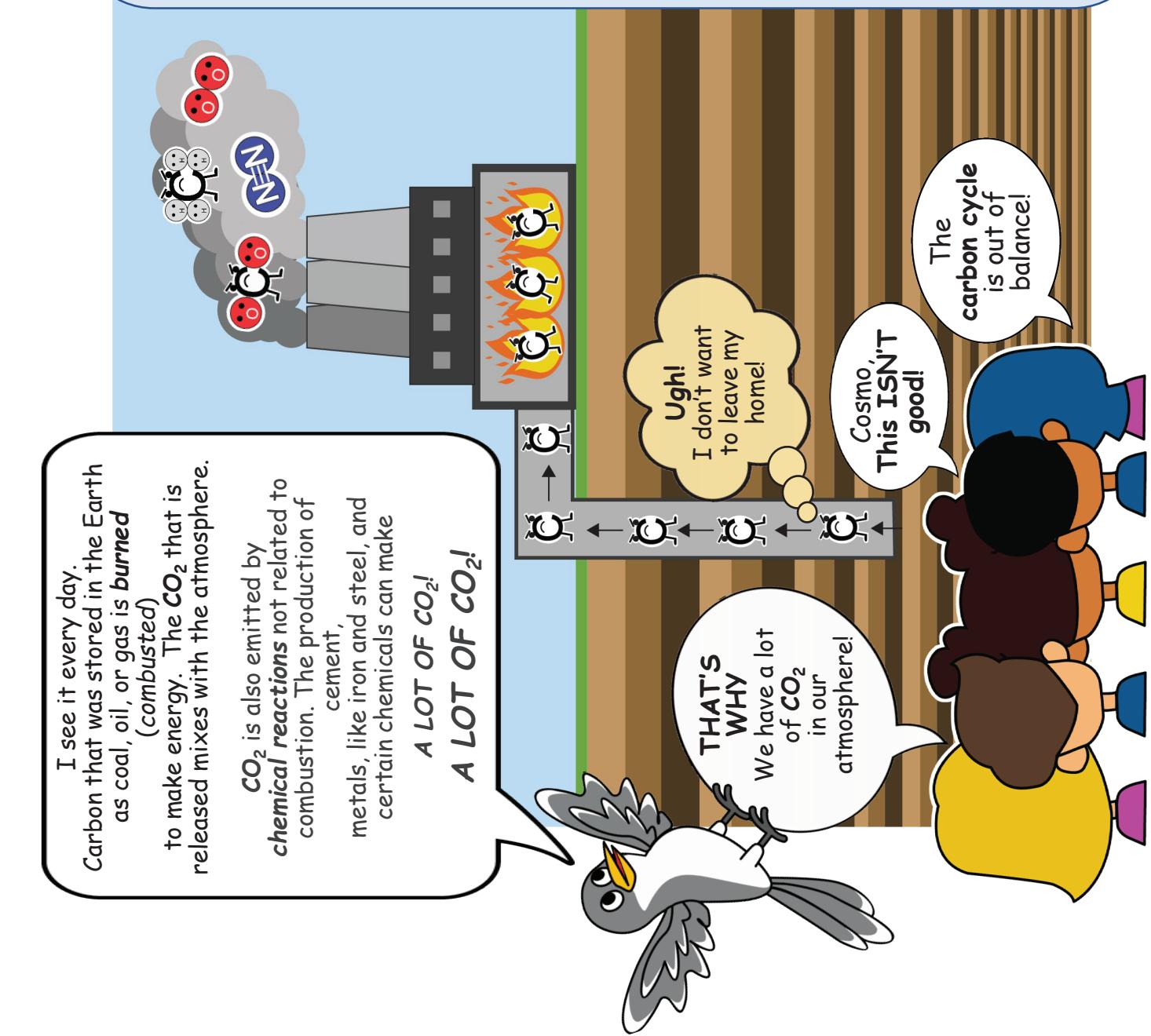
PEOPLE,  
can you stop  
letting CO<sub>2</sub> go?  
THIS has to STOP!  
HAS TO STOP!





I see it every day. Carbon that was stored in the Earth as coal, oil, or gas is **burned** (combusted) to make energy. The **CO<sub>2</sub>** that is released mixes with the atmosphere. **CO<sub>2</sub>** is also emitted by **chemical reactions** not related to combustion. The production of cement, metals, like iron and steel, and certain chemicals can make

**A LOT OF CO<sub>2</sub>!  
A LOT OF CO<sub>2</sub>!**



I'm a chemical element known as **carbon (C)**, and I am the backbone for life.

On Earth, I am mostly stored in rocks and sediments underground.

Fossil fuels contain mostly **C** and **Hydrogen (H)**. **When fossil fuels are burned or used in industrial processes** **C** bonds with O<sub>2</sub> to form CO<sub>2</sub>.



I'm an **oxygen molecule (O<sub>2</sub>)**, and I am essential for living organisms. When burning, I react with C to form CO<sub>2</sub>.



I'm **CO<sub>2</sub>**, a gas that forms when C and O<sub>2</sub> bond. When high volumes of CO<sub>2</sub> accumulate in the atmosphere, heat becomes trapped there.



I'm a **nitrogen molecule (N<sub>2</sub>)**, and I am inert, making me chemically inactive.



I'm a gas known as **methane**.

I am a molecule that contains 1 **C** and 4 **hydrogen atoms**.

I am the 2<sup>nd</sup> most abundant greenhouse gas after CO<sub>2</sub> in industrial waste.

